## CITY COLLEGE B.SC Semester 1 Internal Assessment (online), under CU 2020-21

## CHEMISTRY- GENERAL Paper: CC1/GE 1 (Paper 1) Full Marks – 10

## Attempt all the questions

Q.1. Which one is Amphoteric substance?			
a) Zn	b) Al	c) Both a) and b)	d) none of these.
Q.2. $2HI \rightarrow H_2 + I_2$ , what is the order of the reaction?			
a) 1	b) 2	c) 3	d) 4
Q.3. The $t_{1/2}$ of a first order reaction is 15 min, what is the rate constant of that reaction?			
a) 0.0462 min <sup>-1</sup>	b) 0.0562 mir	n <sup>-1</sup> c) 0.0662 min	d) none of these.
Q.4. What is the electronic configuration of $Fe^{3+}$ ion			
a) [ <i>Ar</i> ]3d <sup>5</sup>	b)[Ar]3d <sup>6</sup> 4s	$c)[Ar]3d^{6}$	d) none of these
Q.5. What is the value of azimuthal quantum no.(1) for f-orbitals			
a) l=3	b) l= 4	c) l= 2	d)l=1
Q.6. The correct order of decreasing stability is			
a) $CH_3C^+HCH_3 > CH_3C^+HOCH_3 > CH_3C^+HCOCH_3$			
b) $CH_3C^+HCH_3 > CH_3C^+HCOCH_3 > CH_3C^+HOCH_3$			
c) $CH_3C^+HCOCH_3 > CH_3C^+HOCH_3 > CH_3C^+HCH_3$			
d) $CH_{3}C^{+}HOCH_{3} > CH_{3}C^{+}HCOCH_{3} > CH_{3}C^{+}HCH_{3}$			
Q.7. which of the following has the highest nucleophilicity in Dimethyl Sulfoxide?			
a) Cl <sup>-</sup>			

- b) F-
- c) I⁻
- d) Br-

Q.8. Which of the following groups has the highest priority in the CIP sequence rules?

a) CH<sub>2</sub>CH<sub>3</sub>

b) CH=CH<sub>2</sub>

c) OCH<sub>3</sub>

d) OH

Q.9. Which of the following compounds displays optical isomerism?
(a) CH<sub>2</sub>(OH)-CH<sub>2</sub>(OH)
(b) CH<sub>3</sub>-CHCl- CH<sub>2</sub>(OH)
(c) CH<sub>3</sub>-CHCl- CH<sub>3</sub>
(d) CH<sub>3</sub>-O-C<sub>2</sub>H<sub>5</sub>

Q.10. Number of hyperconjugation structures in isopropyl radical is

(a) 9

(b) 3

(c) 6

(d) 5